

Asymmetric total synthesis of 9-methoxystrobilurin K

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Abstract—Asymmetric total synthesis of a potent antifungal and cytostatic 9-methoxystrobilurin K was achieved by developing a convergent and versatile synthetic route. © 2001 Elsevier Science Ltd. All rights reserved.

9-Methoxystrobilurin K (1) was first isolated from a mycelial culture of the *Favolaschia* by Anke and coworkers in 1995¹ as a new and potent analogue of β-methoxyacrylate antibiotics (MOAs). The compound was originally proposed as another structure;¹ however, Blunt and Munro et al. claimed a benzodioxepin-type structure 1 in 1997.² Recently, Anke and Steglich et al. formally revised the structure to 1 by the comparison of the synthetic benzodioxepin moiety with the degradation product of natural strobilurin K.³

9-Methoxystrobilurin K (1)

9-Methoxystrobilurin K (1) exhibits strong antifungal activities toward a wide variety of fungi by specific binding to the Qo-center of cytochrome b and inhibiting mitochondrial respiration in a similar manner as other natural and artificial MOAs. In addition, 1 shows remarkable cytostatic activity toward human-derived tumor cell lines at very low concentration without showing any significant cytotoxity. The interesting cytostatic property of 9-methoxystrobilurin K would be closely related to the structure of an extended sidechain moiety from the 1,5-benzodioxepin ring system because strobilurin D and E having similar part on aromatic ring also exhibit cytostatic activities.⁴ Study of

In our previous paper, a total synthesis of 9-methoxy-strobilurin A (2), which is the simplest 9-methoxy-strobilurin-type MOA, was reported.⁵ In this communication, we would like to describe the first total synthesis

9-Methoxystrobilurin A (2)

of the more complicated and potent analogue 9-methoxystrobilurin K by a convergent and applicable method including efficient construction of an enantiomerically pure 1,5-benzodioxepin ring system having a hindered and acid-sensitive allylic tertiary-secondary ether linkage.

Our synthetic strategy for the benzodioxepin-type 9-methoxystrobilurin analogue is shown in Scheme 1. A convergent synthetic route was strongly desired for future studies of the structure—activity relationships of the side-chain moiety. A key-step of our approach is the Heck reaction of aryl bromide with vinyl ketone, and optically active 7-bromo-4,4-dimethyl-3,4-dihydro-1,5-benzodioxepin-3-ol 6 was decided as the first synthetic target. The 1,5-benzodioxepin skeleton in 6 might be constructed by a Lewis acid-mediated intramolecular cyclization of the chiral epoxy phenol 5.

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the structure–activity relationship on the side-chain moiety is expected for the development of new antitumor agents; however, asymmetric synthesis of 9-methoxystrobilurin K or its derivative has not yet been developed.

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Scheme 1. Synthetic strategy for benzodioxepin-type 9-methoxystrobilurins.

The desired chiral epoxy phenol 5 was successfully prepared from prenyl alcohol 3 using Sharpless asymmetric epoxidation (Scheme 2). However, in the synthesis of 7-bromo-type benzodioxepin, the yield of the desired 7-endo-cyclized product 6 significantly decreased because of several undesirable side reactions. The major side reaction was tin(IV) chloride-mediated 1,2-hydride migration of epoxy phenol 5 followed by an intramolecular acetalization resulting in the corresponding hemiketal of isopropyl ketone. This is probably due to the lowering of nucleophilicity of the phenolic hydroxyl group on the bromo-substituted aromatic ring.

Therefore, several reaction conditions for the 7-endo selective cyclization were examined and the yield was greatly improved by use of diethyl ether as solvent at 0°C. Interestingly, a major by-product under these conditions was a chlorohydrin 7 produced by ring-opening and chlorine-transfer from tin(IV) chloride. It is noted that the chlorohydrin 7 was efficiently converted in high yield to the starting epoxy phenol 5 by a treatment with potassium *tert*-butoxide at 0°C.

Next, construction of a hindered and allylic secondarytertiary ether linkage was studied. Several attempts for direct ether synthesis utilizing 1,5-benzodioxepin-3-ol 6 and isoprene under acidic conditions were unsuccessful because of exclusive polymerization of isoprene; therefore, a stepwise approach was attempted. A Williamson-type coupling reaction of methyl 2-bromopropionate with sodium alkoxide of 6 smoothly proceeded to give the corresponding α -alkoxyester 8 in good yield. The thus-obtained α -monomethylester 8 was further methylated with sodium bis(trimethylsilyl)amide-methyl iodide to afford the corresponding α,α -dimethylester 9 and the desired secondary-tertiary ether linkage was successfully constructed. The ester 9 was converted to the corresponding aldehyde 10, and the following Wittig reaction gave the 7-bromo-4,4-dimethyl-1,5-benzodioxepin 11 equipped with the natural-type side-chain moiety (Scheme 3).

Heck reaction of the above 7-bromo-4,4-dimethyl-1,5-benzodioxepin 11 with 2 equivalents of vinyl ketone 12^5 prepared from methyl itaconate was tried to obtain the corresponding α,β -unsaturated ketone 13 (Scheme 4). The reaction proceeded smoothly, and the desired coupling product 13 was prepared in high yield (81%). The conversion of enone 13 to the corresponding methyl enol ether 15 was then examined; however, the allylic and tertiary ether linkage was completely cleaved under the acidic condition employed for the synthesis of 9-methoxystrobilurin A.⁵ On the other hand, O-alkyla-

Scheme 2. Synthesis of optically active 7-bromo-1,5-benzodioxepin-3-ol. (i) TBHP, Ti(O'Pr)₄, (+)-DIPT, CH₂Cl₂, -40°C; (ii) p-NO₂BzCl, Et₃N, 0°C; (iii) recryst. from Et₂O; (iv) NaOMe, MeOH, Et₂O, 0°C; (v) TsCl, DMAP, CH₂Cl₂, Et₃N, 0°C; (vi) K₂CO₃, 5-bromo-2-hydroxyacetophenone **3**, DMF, rt; (vii) mCPBA, benzene, 60°C; (viii) DIBAH, THF, -78°C; (ix) SnCl₄, Et₂O, 0°C; (x) KO'Bu, THF, -15°C.

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Scheme 3. Construction of hindered allylic tertiary-secondary ether linkage. (i) NaH, methyl 2-bromopropionate, THF, 0°C; (ii) NaHMDS, MeI, THF, -78°C; (iii) DIBAH, CH₂Cl₂, -78°C; (iv) (COCl)₂, DMSO, Et₃N, CH₂Cl₂, -60°C; (v) Ph₃P=CH₂, DMSO-THF, rt.

Scheme 4. Total synthesis of 9-methoxystrobilurin K. (i) 20 mol% Pd(OAc)₂, PPh₃, Et₃N, 100°C; (ii) aq. NaOH–MeOH, rt then HCl; (iii) KO'Bu, MeI, DMF, -45 to -15°C; (iv) NaH, HCOOMe, rt; (v) K₂CO₃, Me₂SO₄, HCOOMe, rt; (vi) *hv* (λ: 365 nm), acetone–benzene. rt.

tion of enolate under several basic conditions (e.g. potassium bis(trimethylsilyl)amide-methyl triflate) gave complex mixtures. We assumed that an undesirable side reaction occurred after intramolecular lactonization of the enolic hydroxyl group. In order to prevent the intramolecular reaction, a stepwise double methylation was attempted. First, the ester 13 was hydrolyzed to afford the corresponding carboxylic acid 14. A dianion was generated by treatment of the carboxylic acid 14 with potassium tert-butoxide in dimethylformamide at -45°C. By the addition of excess dimethyl sulfate to the dianion at -45 to -15°C, the double methylation proceeded successively and methyl enol ether 15 was obtained in good yield. Treatment of the enol ether 15 with sodium hydride-methyl formate, followed by the addition of dimethyl sulfate and potassium carbonate gave a mixture of the geometrical isomers of 9methoxystrobilurin K (1). The mixture was isomerized twice by irradiation with an ultraviolet lamp (λ : 365 nm) and the desired 1 was successfully obtained (three steps total 41% yield) along with other isomerizationincomplete isomers.⁷ The physical data of the synthetic 9-methoxystrobilurin K⁸ were in good accordance with those of the natural product reported by Anke's group.¹

Thus, the first asymmetric total synthesis of 9-methoxystrobilurin K was successfully achieved. It should be emphasized that the present methodology could be applied to the synthesis of various 9-methoxystrobilurin-type β -MOAs. Further investigation into the structure–activity relationships and development of a new and pharmacologically superior analogue are now in progress.

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- Physical data of synthesized 9-methoxystrobilurin K (1):
 ¹H NMR (δ, 300 MHz, CDCl₃) 1.23 (s, 3H), 1.32 (s, 6H), 1.43 (s, 3H), 1.89 (s, 3H), 3.65 (s, 3H), 3.68 (dd, 1H, *J*=3.3, 7.9 Hz), 3.71 (s, 3H), 3.81 (s, 3H), 3.98 (dd, 1H,

J=7.7, 12.3 Hz), 4.22 (dd, 1H, J=3.3, 12.3 Hz), 5.15 (d, 1H, J=10.6 Hz), 5.17 (d, 1H, J=17.6 Hz), 5.88 (dd, 1H, J=10.8, 17.6 Hz), 6.37 (d, 1H, J=15.8 Hz), 6.63 (d, 1H, J=15.9 Hz), 6.82 (d, 1H, J=8.6 Hz), 6.93 (dd, 1H, J=2.0, 8.8 Hz), 6.94 (d, 1H, J=2.0 Hz), 7.39 (s, 1H); ¹³C NMR (δ, 75.5 MHz, CDCl₃) 16.3, 21.8, 26.3, 26.7, 28.1, 51.5, 59.5 61.8, 71.8, 75.5, 76.1, 81.6, 110.5, 114.4, 118.1, 120.1, 120.3, 121.5, 122.5, 126.8, 132.9, 143.5, 146.4, 150.6, 152.6, 159.4, 168.1; $[\alpha]_D^{24} = -8.87$ (c=0.860, CHCl₃); HRMS calcd for C₂₂H₃₆O₇ (M⁺) 472.2461, found 472.2437.